Received 2 July 2004

Online 31 July 2004

Accepted 16 July 2004

Acta Crystallographica Section E **Structure Reports** Online

ISSN 1600-5368

Craig C. McLauchlan,* Amy N. Varda and Julian R. Giles

Department of Chemistry, Illinois State University, Campus Box 4160, Normal, IL 61790-4160, USA

Correspondence e-mail: mclauchlan@ilstu.edu

Key indicators

Single-crystal X-ray study T = 293 KMean σ (C–C) = 0.003 Å R factor = 0.061 wR factor = 0.125 Data-to-parameter ratio = 14.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The title compound, C₁₀H₁₀N₆, was crystallized from a methylene dichloride solution. Tpm has gained much attention of late as a neutral scorpionate ligand analogue of hydrotris(pyrazol-1-yl)borate. We report here the structure of the free Tpm ligand, which shows a structure typical of a tripodal neutral ligand. The compound crystallizes in a chiral space group despite the fact that it is achiral, ideally $C_{3\nu}$.

Tris(pyrazol-1-yl)methane (Tpm)

Comment

The ligand tris(pyrazol-1-yl)methane (Tpm), (I), was first reported by Trofimenko in 1970. Tpm has gained much attention of late as an isoelectronic neutral scorpionate ligand analogue of hydrotris(pyrazol-1-yl)borate (Tp⁻) (Reger, 1999; Reger & Grattan, 2003). We report here the structure of the free Tpm ligand. The molecule is achiral in solution with idealized $C_{3\nu}$ (3m) symmetry but crystallizes in the space group $P2_12_12_1$ with four molecules in the asymmetric unit. Examination of the atomic coordinates with the program ADDSYM of the PLATON suite of programs (Spek, 2002) did not suggest the presence of any additional symmetry. The loss of symmetry in the solid state is evident in the varying torsion of the pyrazole rings of the four crystallographically unique Tpm molecules (Table 1).



Experimental

The title compound was prepared according to the procedure of Reger et al. (2000). Crystals suitable for study (m.p. 371-375 K) were obtained by slow evaporation of a methylene dichloride solution. Spectroscopic analysis, ¹H NMR (CDCl₃, p.p.m.): 8.437 (s, 1H), 7.673 (d, 3H), 7.579 (d, 3H), 6.370 (t, 3H); ¹³C NMR (CDCl₃, p.p.m.): 142.08, 129.77, 107.57, 83.45.

Crystal data $C_{10}H_{10}N_6$ Mo $K\alpha$ radiation $M_r = 214.24$ Cell parameters from 13354 Orthorhombic, $P2_12_12_1$ a = 10.7882(5) Å $\theta = 2.1 - 24.7^{\circ}$ b = 15.3791(9) Å c = 25.4634 (13) Å $V = 4224.7 (4) \text{ Å}^3$ *Z* = 16 $D_x = 1.347 \text{ Mg m}^{-3}$

© 2004 International Union of Crystallography Printed in Great Britain - all rights reserved

reflections $\mu = 0.09~\mathrm{mm}^{-1}$ T = 293 (2) KPrism, colorless $0.24 \times 0.10 \times 0.05 \text{ mm}$

organic papers

Data collection

Oxford Diffraction Xcalibur3			
diffractometer equipped with a			
CCD detector			
φ and ω scans			
Absorption correction: numerical			
(Clark & Reid, 1995)			
$T_{\min} = 0.979, \ T_{\max} = 0.992$			
42106 measured reflections			

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.061$ $wR(F^2) = 0.126$ S = 1.138376 reflections 578 parameters

Table 1

Selected torsion angles (°).

C10-N1-N4-C1	170.2 (2)	C30-N13-N16-C21	175.7 (2)
C10-N2-N5-C4	-179.8(2)	C30-N14-N17-C24	172.31 (19)
C10-N3-N6-C7	178.15 (19)	C30-N15-N18-C27	-173.08 (19)
C20-N7-N10-C13	-178.4(2)	C40-N19-N22-C31	176.53 (19)
C20-N8-N11-C14	-172.51(19)	C40-N20-N23-C34	-175.1(2)
C20-N9-N12-C17	-174.6(2)	C40-N21-N24-C37	-174.26 (19)

8376 independent reflections 6743 reflections with $I > 2\sigma(I)$

H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.0568P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} = 0.002$

 $\Delta \rho_{\rm max} = 0.23 \ {\rm e} \ {\rm \AA}^{-3}$

 $\Delta \rho_{\rm min} = -0.25 \text{ e} \text{ Å}^{-3}$

 $\begin{aligned} R_{\text{int}} &= 0.056\\ \theta_{\text{max}} &= 33.5^{\circ}\\ h &= -16 \rightarrow 9\\ k &= -21 \rightarrow 22\\ l &= -37 \rightarrow 38 \end{aligned}$

The value of the absolute structure parameter (Flack, 1983) is meaningless for the title compound as Mo K α radiation was employed and no atoms heavier than N are present. Friedel pair reflections were merged before the final refinement. H atoms were positioned geometrically (C-H = 0.93–0.98 Å) and refined using a riding model, with $U_{iso}(H) = 1.2U_{iso}(C)$.

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2004); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2004); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).





Displacement ellipsoid plot of the asymmetric unit of (I), showing the four crystallographically unique molecules of Tpm.

This research was supported by Illinois State University. JRG acknowledges the support of Project SEED of the American Chemical Society. CCM thanks Dr A. A. Narducci Sarjeant of Johns Hopkins University for experimental assistance and helpful discussion.

References

- Clark, R. C. & Reid, J. S. (1995). Acta Cryst. A51, 887-897.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Flack, H. D. (1983). Acta Cryst. A39, 876-881.
- Oxford Diffraction (2004). CrysAlis CCD and CrysAlis RED. Versions 1.171.21 beta. Oxford Diffraction Ltd, Abingdon, Oxfordshire, England.
- Reger, D. L. (1999). Comments Inorg. Chem. 21, 1–28.
- Reger, D. L. & Grattan, T. C. (2003). Synthesis, 3, 350–356.
- Reger, D. L., Grattan, T. C., Brown, K. J., Little, C. A., Lamba, J. J. S., Rheingold, A. L. & Sommer, R. D. (2000). J. Organomet. Chem. 607, 120– 128.
- Sheldrick, G. M. (1997). *SHELXS* and *SHELXL*97. University of Göttingen, Germany.
- Spek, A. L. (2002). *PLATON*. Version of August 2002. University of Utrecht, The Netherlands.
- Trofimenko, S. (1970). J. Am. Chem. Soc. 92, 5118-5126.